

# Thermodynamics Problems And Answers

This eminently readable introductory text provides a sound foundation to understand the abstract concepts used to express the laws of thermodynamics. The emphasis is on the fundamentals rather than spoon-feeding the subject matter. The concepts are explained with utmost clarity in simple and elegant language. It provides the background material needed for students to solve practical problems related to thermodynamics. Answers to all problems are provided.

A comprehensive, best-selling introduction to the basics of engineering thermodynamics. Requiring only college-level physics and calculus, this popular book includes a realistic art program to give more realism to engineering devices and systems. A tested and proven problem-solving methodology encourages readers to think systematically and develop an orderly approach to problem solving: Provides readers with a state-of-the art introduction to second law analysis. Design/open-ended problems provide readers with brief design experiences that offer them opportunities to apply constraints and consider alternatives.

Problems and Solutions on Thermodynamics and Statistical Mechanics World Scientific

This book is a very useful reference that contains worked-out solutions for all the exercise problems in the book Chemical Engineering Thermodynamics by the same author. Step-by-step solutions to all exercise problems are provided and solutions are explained with detailed

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and extensive illustrations. It will come in handy for all teachers and users of Chemical Engineering Thermodynamics.

Solutions to Selected Problems In a Course in Statistical Thermodynamics is the companion book to A Course in Statistical Thermodynamics. This title provides the solutions to a select number of problems contained in the main title. The problem sets explore the physical aspects of the methodology of statistical thermodynamics without the use of advanced mathematical methods. This book is divided into 14 chapters that focus on such items as the statistical method to various specialized applications of statistical thermodynamics.

Heat and Thermodynamics is meant for an introductory course on Heat and Thermodynamics. Emphasis has been given to the fundamentals of thermodynamics. The book uses variety of diagrams, charts and learning aids to enable easy understanding of the s

A companion book to the textbook The Exergy Method of Thermal Plant Analysis. This Companion Book presents model solutions to the questions taken from Appendix G of the main textbook. Since the Exergy Method is a relatively new area of Applied Thermodynamics it was thought that the presentation of model solutions of problems of various types would be of some help both to teachers and to self-teaching students. The advantages of the use of exergy analysis were demonstrated by pointing out and quantifying thermodynamic losses of various plant components and plant configurations. These were discussed at the end of the solutions under

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Comments. It is hoped that this will give students a deeper understanding of the nature of irreversibilities of various kinds and their effect on plant performance. Dr Tadeusz J. Kotas joined the Department of Mechanical Engineering of Queen Mary College as a member of teaching staff in 1957. His main areas of interest were Mechanics of Fluids and Applied Thermodynamics, obtaining a PhD degree for his work in the former subject. His work in the latter subject focused on the Exergy Method, contributing to its development through his research and publications and to its dissemination through courses which he ran in Britain and in a number of European countries for practicing engineers and academics.

Take some heat off the complexity of thermodynamics Does the mere thought of thermodynamics make you sweat? It doesn't have to! This hands-on guide helps you score your highest in a thermodynamics course by offering easily understood, plain-English explanations of how energy is used in things like automobiles, airplanes, air conditioners, and electric powerplants.

Thermodynamics 101— take a look at some examples of both natural and man-made thermodynamic systems and get a handle on how energy can be used to perform work Turn up the heat — discover how to use the first and second laws of thermodynamics to determine (and improve upon) the efficiency of machines Oh, behave — get the 411 on how gases behave and relate to one another in different situations, from ideal-gas laws to real gases Burn with desire — find out everything you need to know about conserving mass and energy in

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combustion processes Open the book and find: The laws of thermodynamics Important properties and their relationships The lowdown on solids, liquids, and gases How work and heat go hand in hand The cycles that power thermodynamic processes Chemical mixtures and reactions Ten pioneers in thermodynamics Real-world applications of thermodynamic laws and concepts Learn to: Master the concepts and principles of thermodynamics Develop the problem-solving skills used by professional engineers Ace your thermodynamics course

Beyond Equilibrium Thermodynamics fills a niche in the market by providing a comprehensive introduction to a new, emerging topic in the field. The importance of non-equilibrium thermodynamics is addressed in order to fully understand how a system works, whether it is in a biological system like the brain or a system that develops plastic. In order to fully grasp the subject, the book clearly explains the physical concepts and mathematics involved, as well as presenting problems and solutions; over 200 exercises and answers are included.

Engineers, scientists, and applied mathematicians can all use the book to address their problems in modelling, calculating, and understanding dynamic responses of materials.

This is a review book for people planning to take the PE exam in Chemical Engineering. Prepared specifically for the exam used in all 50 states. It features 188 new PE problems with detailed step by step solutions. The book covers all topics on the exam, and includes easy to use tables, charts, and formulas. It is an ideal desk

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Companion to DAS's Chemical Engineer License Review. It includes sixteen chapters and a short PE sample exam as well as complete references and an index. Chapters include the following topical areas: material and energy balances; fluid dynamics; heat transfer; evaporation; distillation; absorption; leaching; liq-liq extraction; psychrometry and humidification, drying, filtration, thermodynamics, chemical kinetics, process control, mass transfer, and plant safety. The ideal study guide, this book brings all elements of professional problem solving together in one BIG BOOK. Ideal desk reference. Answers hundreds of the most frequently asked questions. The first truly practical, no-nonsense problems and solution book for the difficult PE exam. Full step-by-step solutions are included.

Problems in Metallurgical Thermodynamics and Kinetics provides an illustration of the calculations encountered in the study of metallurgical thermodynamics and kinetics, focusing on theoretical concepts and practical applications. The chapters of this book provide comprehensive account of the theories, including basic and applied numerical examples with solutions.

Unsolved numerical examples drawn from a wide range of metallurgical processes are also provided at the end of each chapter. The topics discussed include the three laws of thermodynamics; Clausius-Clapeyron equation; fugacity, activity, and equilibrium constant; thermodynamics of electrochemical cells; and kinetics. This book is beneficial to undergraduate and postgraduate students in universities, polytechnics, and technical colleges.

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Modern Engineering Thermodynamics is designed for use in a standard two-semester engineering thermodynamics course sequence. The first half of the text contains material suitable for a basic Thermodynamics course taken by engineers from all majors. The second half of the text is suitable for an Applied Thermodynamics course in mechanical engineering programs. The text has numerous features that are unique among engineering textbooks, including historical vignettes, critical thinking boxes, and case studies. All are designed to bring real engineering applications into a subject that can be somewhat abstract and mathematical. Over 200 worked examples and more than 1,300 end of chapter problems provide opportunities to practice solving problems related to concepts in the text. Provides the reader with clear presentations of the fundamental principles of basic and applied engineering thermodynamics. Helps students develop engineering problem solving skills through the use of structured problem-solving techniques. Introduces the Second Law of Thermodynamics through a basic entropy concept, providing students a more intuitive understanding of this key course topic. Covers Property Values before the First Law of Thermodynamics to ensure students have a firm understanding of property data before using them. Over 200 worked examples and more than 1,300 end of chapter problems offer students extensive opportunity to practice solving problems. Historical Vignettes, Critical Thinking boxes and Case Studies throughout the book help relate abstract concepts to actual engineering applications. For greater

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instructor flexibility at exam time, thermodynamic tables are provided in a separate accompanying booklet. Available online testing and assessment component helps students assess their knowledge of the topics. Email [textbooks@elsevier.com](mailto:textbooks@elsevier.com) for details.

The laws of thermodynamics have wide ranging practical applications in all branches of engineering. This invaluable textbook covers all the subject matter in a typical undergraduate course in engineering thermodynamics, and uses carefully chosen worked examples and problems to expose students to diverse applications of thermodynamics. This new edition has been revised and updated to include two new chapters on thermodynamic property relations, and the statistical interpretation of entropy. Problems with numerical answers are included at the end of each chapter. As a guide, instructors can use the examples and problems in tutorials, quizzes and examinations. Request Inspection Copy

This book is the solution manual to the textbook "A Modern Course in University Physics". It contains solutions to all the problems in the aforementioned textbook. This solution manual is a good companion to the textbook. In this solution manual, we work out every problem carefully and in detail. With this solution manual used in conjunction with the textbook, the reader can understand and grasp the physics ideas more quickly and deeply. Some of the problems are not purely exercises; they contain extension of the materials covered in the textbook. Some of the problems contain problem-solving techniques that are not covered in the

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textbook. Request Inspection Copy

This inter-disciplinary guide to the thermodynamics of living organisms has been thoroughly revised and updated to provide a uniquely integrated overview of the subject. Retaining its highly readable style, it will serve as an introduction to the study of energy transformation in the life sciences and particularly as an accessible means for biology, biochemistry and bioengineering undergraduate students to acquaint themselves with the physical dimension of their subject. The emphasis throughout the text is on understanding basic concepts and developing problem-solving skills. The mathematical difficulty increases gradually by chapter, but no calculus is required. Topics covered include energy and its transformation, the First Law of Thermodynamics, Gibbs free energy, statistical thermodynamics, binding equilibria and reaction kinetics. Each chapter comprises numerous illustrative examples taken from different areas of biochemistry, as well as a broad range of exercises and references for further study.

Here is a comprehensive and comprehensible treatment of engineering thermodynamics from its theoretical foundations to its applications in real situations. The thermodynamics presented will prepare students for later courses in fluid mechanics and heat transfer, and practicing engineers will find the applications helpful in their professional work. The book is appropriate for an introductory undergraduate course in thermodynamics and for a subsequent course in thermodynamic applications. The chapters dealing with steam power plants, internal combustion engines, and HVAC are

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unmatched. The introductory chapter on turbomachinery is also unique. A thorough development of the second law of thermodynamics is provided in chapters 7-9. The ramifications of the second law receive thorough discussion; the student not only performs calculations, but understands the implications of the calculated results. Computer models created in TK Solver accompany each chapter and are particularly useful in the application areas. The TK Solver files provided with the book can be used as written or modified and merged into models developed to analyze new problems. The book has two particularly important strengths: its readability and the depth of its treatment of applications. The readability will make the content understandable to the average students; the depth in applications will make the book suitable for applied upper-level courses as well. This introductory textbook for standard undergraduate courses in thermodynamics has been completely rewritten to explore a greater number of topics, more clearly and concisely. Starting with an overview of important quantum behaviours, the book teaches students how to calculate probabilities in order to provide a firm foundation for later chapters. It introduces the ideas of classical thermodynamics and explores them both in general and as they are applied to specific processes and interactions. The remainder of the book deals with statistical mechanics. Each topic ends with a boxed summary of ideas and results, and every chapter contains numerous homework problems, covering a broad range of difficulties. Answers are given to odd-numbered problems, and solutions to even-numbered

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problems are available to instructors at [www.cambridge.org/9781107694927](http://www.cambridge.org/9781107694927).

This textbook on thermodynamics is intended primarily for honours and B. Sc students majoring in physical chemistry. However, students of physics, engineering and biochemistry will also find the book relevant to their studies. Its principal features are a much shorter presentation of the laws of thermodynamics than is customary, made possible by the definition of the thermodynamic scale of temperature using only one fixed point (the triple point of water) which immediately follows the Zeroth Law. The author's first exposure to thermodynamics revealed that its usefulness seemed to be mostly confined to the study of gases in equilibrium. Readers of this book will find that applications of thermodynamics to liquids and solids as well as gases are emphasized, and they will learn that thermodynamics can be applied to systems which are not in equilibrium. This book contains three learning aids. Fully worked out examples are included at appropriate places in the text, which also includes numerous exercises. These are designed to help the reader stop and think about what he or she has just read. Answers to the exercises are given at the end of each section and there are also problems at the end of each chapter which readers can work out on their own.

This book contains a modern selection of about 200 solved problems and examples arranged in a didactic way for hands-on experience with course work in a standard advanced undergraduate/first-year graduate class in thermodynamics and statistical physics. The

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principles of thermodynamics and equilibrium statistical physics are few and simple, but their application often proves more involved than it may seem at first sight. This book is a comprehensive complement to any textbook in the field, emphasizing the analogies between the different systems, and paves the way for an in-depth study of solid state physics, soft matter physics, and field theory.

Chemical engineers face the challenge of learning the difficult concept and application of entropy and the 2nd Law of Thermodynamics. By following a visual approach and offering qualitative discussions of the role of molecular interactions, Koretsky helps them understand and visualize thermodynamics. Highlighted examples show how the material is applied in the real world.

Expanded coverage includes biological content and examples, the Equation of State approach for both liquid and vapor phases in VLE, and the practical side of the 2nd Law. Engineers will then be able to use this resource as the basis for more advanced concepts.

REA's Thermodynamics Problem Solver Each Problem Solver is an insightful and essential study and solution guide chock-full of clear, concise problem-solving gems. Answers to all of your questions can be found in one convenient source from one of the most trusted names in reference solution guides. More useful, more practical, and more informative, these study aids are the best review books and textbook companions available.

They're perfect for undergraduate and graduate studies. This highly useful reference provides thorough coverage of pressure, work and heat, energy, entropy, first and

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second laws, ideal gas processes, vapor refrigeration cycles, mixtures, and solutions. For students in engineering, physics, and chemistry.

Designed for use in a standard two-semester engineering thermodynamics course sequence. The first half of the text contains material suitable for a basic Thermodynamics course taken by engineers from all majors. The second half of the text is suitable for an Applied Thermodynamics course in mechanical engineering programs. The text has numerous features that are unique among engineering textbooks, including historical vignettes, critical thinking boxes, and case studies. All are designed to bring real engineering applications into a subject that can be somewhat abstract and mathematical. Over 200 worked examples and more than 1,300 end of chapter problems provide the use opportunities to practice solving problems related to concepts in the text. Provides the reader with clear presentations of the fundamental principles of basic and applied engineering thermodynamics. Helps students develop engineering problem solving skills through the use of structured problem-solving techniques. Introduces the Second Law of Thermodynamics through a basic entropy concept, providing students a more intuitive understanding of this key course topic. Covers Property Values before the First Law of Thermodynamics to ensure students have a firm understanding of property data before using them. Over 200 worked examples and more than 1,300 end of chapter problems offer students extensive opportunity to practice solving problems. Historical Vignettes, Critical Thinking boxes and Case Studies throughout the book help relate abstract concepts to actual engineering applications. For greater instructor flexibility at exam time, thermodynamic tables are provided in a separate accompanying booklet. Available

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online testing and assessment component helps students assess their knowledge of the topics. Email [textbooks@elsevier.com](mailto:textbooks@elsevier.com) for details.

This volume presents a sound foundation for understanding abstract concepts (physical properties such as fugacity, or chemical processes, such as distillation) of phase and reaction equilibria, and shows you how to apply these concepts to solve practical problems using numerous, clear examples. The book encourages the use of MATHCAD to write programs specific to each problem, enabling you to easily track mistakes and understand the order of magnitude of the various quantities involved. Provides guidelines in order to choose the 'best' equation of state suitable for the particular situation Includes up-to-date information, comprehensive in-depth content and current examples in each chapter Provides the right tools in order to and encourages you to use MATHCAD to write your own specific programs Includes many well organized problems (with solutions), which are extensions of the examples enabling conceptual understanding to quantitative/real problem solving Includes all mathematical background required for solving problems encountered in phase and reaction equilibria Provides a Solutions Manual (for instructors in pdf form) allowing the use of the book in advanced thermodynamic courses

Publisher's Note: Products purchased from Third Party sellers are not guaranteed by the publisher for quality, authenticity, or access to any online entitlements included with the product. Tough Test Questions? Missed Lectures? Not Enough Time? Fortunately, there's Schaum's. More than 40 million students have trusted Schaum's to help them succeed in the classroom and on exams. Schaum's is the key to faster learning and higher grades in every subject. Each Outline presents all the essential course information in

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an easy-to-follow, topic-by-topic format. You also get hundreds of examples, solved problems, and practice exercises to test your skills. Schaum's Outline of Thermodynamics for Engineers, Fourth Edition is packed with four sample tests for the engineering qualifying exam, hundreds of examples, solved problems, and practice exercises to test your skills. This updated guide approaches the subject in a more concise, ordered manner than most standard texts, which are often filled with extraneous material. Schaum's Outline of Thermodynamics for Engineers, Fourth Edition features:

- 889 fully-solved problems
- 4 sample tests for the engineering qualifying exam
- An accessible review of thermodynamics
- Chapter on refrigeration cycles
- Nomenclature reflecting current usage
- Support for all the major leading textbooks in thermodynamics
- Content that is appropriate for Thermodynamics, Engineering Thermodynamics, Principles of Thermodynamics, Fundamentals of Thermodynamics, and Thermodynamics I & II courses

PLUS: Access to the revised Schaums.com website and new app, containing 20 problem-solving videos, and more. Schaum's reinforces the main concepts required in your course and offers hundreds of practice exercises to help you succeed. Use Schaum's to shorten your study time--and get your best test scores! Schaum's Outlines – Problem solved.

Worked Problems in Heat, Thermodynamics and Kinetic Theory for Physics Students is a complementary to textbooks in physics. This book is a collection of exercise problems that have been part of tutorial classes in heat and thermodynamics at the University of London. This collection of exercise problems, with answers that are fully worked out, deals with various topics. This book poses problems covering the definition of temperature such as calculating the assigned value of the temperature of boiling water under specific

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conditions. This text also gives example of problems dealing with the first law of thermodynamics and with the definition of thermal capacities. Some practical questions such as problems dealing with thermal engines are presented. This book then discusses problems using the energy equation, as well as asking the student to derive a general equation of state of a material satisfying a specific condition. This text challenges the student to use a T-S diagram to calculate the efficiency of a reversible cycle under certain conditions. Several other problems concern the Joule and Joule-Kelvin effects, low temperature physics, and heat conduction. This review material can be helpful for students of physics, thermodynamics, and related subjects. It can also be used by teachers of physics.

Phase Diagrams and Thermodynamic Modeling of Solutions provides readers with an understanding of thermodynamics and phase equilibria that is required to make full and efficient use of these tools. The book systematically discusses phase diagrams of all types, the thermodynamics behind them, their calculations from thermodynamic databases, and the structural models of solutions used in the development of these databases. Featuring examples from a wide range of systems including metals, salts, ceramics, refractories, and concentrated aqueous solutions, Phase Diagrams and Thermodynamic Modeling of Solutions is a vital resource for researchers and developers in materials science, metallurgy, combustion and energy, corrosion engineering, environmental engineering, geology, glass technology, nuclear engineering, and other fields of inorganic chemical and materials science and engineering. Additionally, experts involved in developing thermodynamic databases will find a comprehensive reference text of current solution models. Presents a rigorous and complete development of thermodynamics for readers who already have a basic understanding of chemical

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thermodynamics Provides an in-depth understanding of phase equilibria Includes information that can be used as a text for graduate courses on thermodynamics and phase diagrams, or on solution modeling Covers several types of phase diagrams (paraequilibrium, solidus projections, first-melting projections, Scheil diagrams, enthalpy diagrams), and more

If you want top grades and an excellent understanding of thermodynamics, this powerful study tool is the best tutor you can have! It takes you step by step through the subject, giving you lots of example problems with fully worked solutions. You also get hundreds of additional problems to solve on your own, working at your own speed. This Schaum's Outline of Thermodynamics for Engineers gives you clear explanations of theory, as well as numerous examples of practical applications. And the fully solved problems show you just how to work the kinds of questions you'll face on exams!

This textbook is for a one semester introductory course in thermodynamics, primarily for use in a mechanical or aerospace engineering program, although it could also be used in an engineering science curriculum. The book contains a section on the geometry of curves and surfaces, in order to review those parts of calculus that are needed in thermodynamics for interpolation and in discussing thermodynamic equations of state of simple substances. It presents the First Law of Thermodynamics as an equation for the time rate of change of system energy, the same way that Newton's Law of Motion, an equation for the time rate of change of system momentum, is presented in Dynamics. Moreover, this emphasis illustrates the importance of the equation to the study of heat transfer and fluid mechanics. New thermodynamic properties, such as internal energy and entropy, are introduced with a motivating discussion rather than by abstract postulation, and connection is made with

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kinetic theory. Thermodynamic properties of the vaporizable liquids needed for the solution of practical thermodynamic problems (e.g. water and various refrigerants) are presented in a unique tabular format that is both simple to understand and easy to use. All theoretical discussions throughout the book are accompanied by worked examples illustrating their use in practical devices. These examples of the solution of various kinds of thermodynamic problems are all structured in exactly the same way in order to make, as a result of the repetitions, the solution of new problems easier for students to follow, and ultimately, to produce themselves. Many additional problems are provided, half of them with answers, for students to do on their own.

Thermodynamics Problem Solving in Physical Chemistry: Study Guide and Map is an innovative and unique workbook that guides physical chemistry students through the decision-making process to assess a problem situation, create appropriate solutions, and gain confidence through practice solving physical chemistry problems. The workbook includes six major sections with 20 - 30 solved problems in each section that span from easy, single objective questions to difficult, multistep analysis problems. Each section of the workbook contains key points that highlight major features of the topic to remind students of what they need to apply to solve problems in the topic area. Key Features: Includes a visual map that shows how all the “equations” used in thermodynamics are connected and how they are derived from the three major energy laws. Acts as a guide in deriving the correct solution to a problem. Illustrates the questions students should ask themselves about the critical features of the concepts to solve

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problems in physical chemistry Can be used as a stand-alone product for review of Thermodynamics questions for major tests.

"This textbook addresses the key questions in both classical thermodynamics and statistical thermodynamics: Why are the thermodynamic properties of a nano-sized system different from those of a macroscopic system of the same substance? Why and how is entropy defined in thermodynamics, and how is the entropy change calculated when dissipative heat is involved? What is an ensemble and why is its theory so successful?" "Translated from a highly successful Chinese book, this expanded English edition contains many updated sections and several new ones. They include the introduction of the grand canonical ensemble, the grand partition function and its application to ideal quantum gases, a discussion of the mean field theory of the Ising model and the phenomenon of ferromagnetism, as well as a more detailed discussion of ideal quantum gases near  $T = 0$ , for both Fermi and Bose gases."--BOOK JACKET.

A Practical, Up-to-Date Introduction to Applied Thermodynamics, Including Coverage of Process Simulation Models and an Introduction to Biological Systems Introductory Chemical Engineering Thermodynamics, Second Edition, helps readers master the fundamentals of applied thermodynamics as practiced today: with extensive development of molecular perspectives that enables adaptation to fields including biological systems, environmental applications, and nanotechnology. This text is distinctive in making

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molecular perspectives accessible at the introductory level and connecting properties with practical implications. Features of the second edition include Hierarchical instruction with increasing levels of detail: Content requiring deeper levels of theory is clearly delineated in separate sections and chapters Early introduction to the overall perspective of composite systems like distillation columns, reactive processes, and biological systems Learning objectives, problem-solving strategies for energy balances and phase equilibria, chapter summaries, and “important equations” for every chapter Extensive practical examples, especially coverage of non-ideal mixtures, which include water contamination via hydrocarbons, polymer blending/recycling, oxygenated fuels, hydrogen bonding, osmotic pressure, electrolyte solutions, zwitterions and biological molecules, and other contemporary issues Supporting software in formats for both MATLAB® and spreadsheets Online supplemental sections and resources including instructor slides, ConcepTests, coursecast videos, and other useful resources

Volume 5.

This is a textbook for the standard undergraduate-level course in thermal physics. The book explores applications to engineering, chemistry, biology, geology, atmospheric science, astrophysics, cosmology, and everyday life.

Updated and enhanced with numerous worked-out examples and exercises, this Second Edition continues to present a thorough, concise and accurate discussion of fundamentals and principles of thermodynamics. It

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focuses on practical applications of theory and equips students with sound techniques for solving engineering problems. The treatment of the subject matter emphasizes the phenomena which are associated with the various thermodynamic processes. The topics covered are supported by an extensive set of example problems to enhance the student's understanding of the concepts introduced. The end-of-chapter problems serve to aid the learning process, and extend the material covered in the text by including problems characteristic of engineering design. The book is designed to serve as a text for undergraduate engineering students for a course in thermodynamics.

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